LIFETIME ASSESSMENT OF RADIANT BURNERS MADE OF SHS-INTERMETALLICS*

A. MAZNOY

Tomsk Scientific Center SB RAS, 10/4 Akademicheskii pr., Tomsk, 634055, Russia, maznoy_a@mail.ru, +79234124765

During the last decade, thin-shell radiant burners [1] made of porous SHS-intermetallics [2] have been developing in TSC SB RAS. Low oxidation resistance is a barrier to the successful commercialization of the burners. My presentation will be focused on recent results and prospects for the high-temperature oxidation resistance of porous Ni-Al-Cr intermetallics. The first results have shown that porous SHS-intermetallics with spheroidal elements of 1-2 mm in diameter can work for one year or more at 1000-1100 °C (Fig.1).

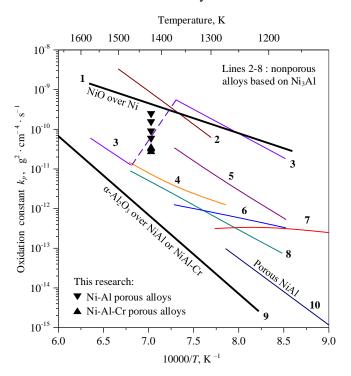


Fig.1. Arrhenius diagrams of k_P vs 10000/T for oxidation of SHS-porous alloys. Lines are literature data: 1 - NiO scale over Ni [3], $2 - \text{pure Ni}_3\text{Al}$ [4], 3 - Ni-11.8wt.%Al alloy [3], $4 - \text{pure Ni}_3\text{Al}$ [5], 5 - IC221M alloy (Ni₃Al + Cr-Zr-Mo-B) [6], $6 - \text{nanocrystaline Ni}_3\text{Al}$ made by MA & SPS [7], $7 - \text{powder-metallurgical Ni}_3\text{Al}$ [8], $8 - \text{pure Ni}_3\text{Al}$ [3], $9 - \alpha - \text{Al}_2\text{O}_3$ scale over B2-NiAl or NiAl-Cr [9], 10 - porous NiAl [10].

REFERENCES

- [1] A. Maznoy, N. Pichugin, I. Yakovlev, R. Fursenko, D. Petrov, S.S. Shy, "Fuel Interchangeability for Lean Premixed Combustion in Cylindrical Radiant Burner Operated in the Internal Combustion Mode," Appl. Therm. Eng. vol. 186, no. 115997. March 2021.
- [2] A. Maznoy, A. Kirdyashkin, V. Kitler, N. Pichugin, V. Salamatov, K. Tcoi, "Self-propagating high-temperature synthesis of macroporous B2+L1₂ Ni-Al intermetallics used in cylindrical radiant burners," J. Alloys Compd. vol. 792, pp. 561–573, July 2019.
- [3] J.D. Kuenzly, D.L. Douglass, "The oxidation mechanism of Ni₃Al containing yttrium," vol. 8, pp. 139–178. June 1974.
- [4] S. Taniguchi, T. Shibata, H. Tsuruoka, "Isothermal oxidation behavior of Ni₃Al-0.1B base alloys containing Ti, Zr, or Hf additions," Oxid. Met. 1986 261. vol. 26, pp. 1–17. August 1986.
- [5] S.C. Choi, H.J. Cho, Y.J. Kim, D.B. Lee, "High-temperature oxidation behavior of pure Ni₃Al," Oxid. Met. vol. 46, pp. 51–72. August 1996.
- [6] D.B. Lee, M.L. Santella, "High temperature oxidation of Ni₃Al alloy containing Cr, Zr, Mo, and B," Mater. Sci. Eng. A. vol. 374, pp. 217–223. June 2004.
- [7] G. Cao, L. Geng, Z. Zheng, M. Naka, "The oxidation of nanocrystalline Ni3Al fabricated by mechanical alloying and spark plasma sintering," Intermetallics. vol. 15, pp. 1672–1677. December 2007.
- [8] P. Pérez, J.L. González-Carrasco, P. Adeva, "Oxidation behavior of a Ni₃Al PM alloy," Oxid. Met. vol. 48, pp. 143–170. August 1997.
- [9] M.W. Brumm, H.J. Grabke, "The oxidation behaviour of NiAl-I. Phase transformations in the alumina scale during oxidation of NiAl and NiAl-Cr alloys," Corros. Sci. vol. 33, pp. 1677–1690. November 1992.
- [10] H.X. Dong, Y. Jiang, Y.H. He, J. Zou, N.P. Xu, B.Y. Huang, C.T. Liu, P.K. Liaw, "Oxidation behavior of porous NiAl prepared through reactive synthesis," Mater. Chem. Phys. vol. 122, pp. 417–423. August 2010.

_

^{*} The research was funded by Russian Science Foundation (project № 21-79-10445).